



DFT, molecular docking and molecular dynamics simulation studies on natural chromone derivatives from *Cassia nomame* for their possible antiviral activity against Coronavirus, SARS-CoV-2

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Abstract

In this study, two naturally occurred chromone derivatives obtained from *Cassia nomame* which are recently entered the literature, have been investigated computationally for their potential antiviral activity against SARS-CoV-2. In the first part of the study, DFT calculations were performed on the investigated compounds. In this part, geometry optimizations, frequency analyses, molecular electrostatic potential map calculations, frontier molecular orbital calculations and NMR spectral studies have been performed. In the second part of the study, molecular docking calculations were performed. SARS-CoV-2 main protease (SARS-CoV-2 M^{pro}) was selected as receptor for molecular docking calculations. In the third part of the study, molecular dynamics simulation studies were performed on the top scoring SARS-CoV-2 M^{pro} – ligand complexes. In this part, binding free energy calculations were also performed on the SARS-CoV-2 M^{pro} – ligand complexes with the use of molecular mechanics with Poisson-Boltzmann surface area (MM-PBSA) method. Results showed that, two naturally occurred chromone derivatives, 5-(isobutyryl)-2-(2-oxopropyl)-7-methoxy-4*H*-chromen-4-one and 5-(isobutyryl)-2-(2-oxopropyl)-6-methoxy-4*H*-chromen-4-one, showed quite high binding affinity to SARS-CoV-2 M^{pro} and remained stable during the molecular dynamics simulations. Additionally, in the last part of the study, drug-likeness analyses were performed on the investigated compounds with the use of Lipinski's rule of five and no violation was observed.

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1. Introduction

Since its first appearance in China in December 2019, SARS-CoV-2 has been causing human pulmonary infections and despite all the measures taken, it continues to spread all over the world very quickly. Currently, vaccination is the only effective way to treat COVID-19, and unfortunately, a specific drug treatment has not been developed yet, and it is a critical issue to propound specific drug treatment options today and in near future.

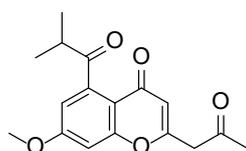
Drug treatment investigations for SARS-CoV-2 infections can be divided into two groups: (1) To use the drugs currently used in SARS and MERS treatments due to the similarity between the target proteins and (2) to develop new specific drugs. Currently, intensive studies are being conducted on whether drugs used in the treatment of SARS and MERS can be used against SARS-CoV-2 such as favipiravir, ribavirin, remdesivir, galidesivir,

disulfiram, lopinavir, ritonavir etc. and other synthesized or isolated compounds found in databases are being investigated against SARS-CoV-2. [1–8] Since it has an important role in viral replication and transcription, SARS-CoV-2 M^{pro} is a commonly selected target in drug repurposing and new drug development studies for the treatment of COVID-19. [9,10]

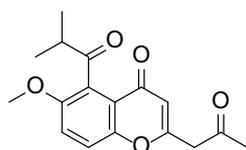
In this study, two novel natural products which were isolated by Liao et al. [11] from *Cassia nomame* have been investigated computationally for their possible antiviral activity against SARS-CoV-2 via performing molecular docking calculations and molecular dynamics simulations. The structures of these natural chromone derivatives are given in Figure 1.

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In the first part of the study, density functional theory calculations were performed on the investigated compounds.



5-(isobutyryl)-2-(2-oxopropyl)-7-methoxy-4H-chromen-4-one
Comp.1



5-(isobutyryl)-2-(2-oxopropyl)-6-methoxy-4H-chromen-4-one
Comp.2

Figure 1. Chemical structures of Comp. 1 and 2.

Geometry optimizations, frequency analyses, MEP (molecular electrostatic potential) map calculations, frontier molecular orbital (FMO) calculations and NMR spectral studies were carried out. In the second part, molecular docking calculations were performed, and then molecular dynamics simulations were carried out on the top-scoring enzyme – ligand complexes, and binding free energies were determined. Finally, drug-likeness analyses were performed on the investigated compounds.

2. Materials and Methods

2.1. DFT calculations

In this part, the title compounds have been investigated computationally and geometry optimizations, frequency analyses, MEP map calculations, FMO calculations and NMR spectral analyses were performed. In DFT calculations, Gaussian 09 Rev.D.01 [12], GaussView 5 [13], VeraChem VConf [14] and Avogadro 1.1.1 [15] software packages were used. All calculations were performed with the use of DFT B3LYP method and various basis sets including 6-31+G(d), 6-31+G(d,p), 6-311+G(d,p) and 6-311+G(2d,p). Geometry optimizations were performed in gas phase. Prior to geometry optimizations, a conformational search was performed for each structure and frequency analyses were also performed to confirm that each optimized geometry corresponds to a global minimum.

2.2. Molecular docking calculations and molecular dynamics simulations

Geometry optimized structures of Comp. 1 and 2 were used for molecular docking calculations. In molecular docking calculations AutoDock Tools [16] and

AutoDock Vina were used and Discovery Studio Visualizer [17] was used for the representation of the docking results. 3D structure of SARS-CoV-2 M^{pro} was obtained from RCSB Protein Data Bank [18,19] (PDB ID:5R80). Prior to molecular docking, water molecules and the bound ligands in the structure of the enzyme were removed, hydrogen atoms and Gasteiger charges were added, and docking calculations were performed with the use of Lamarckian genetic algorithm. After performing molecular docking calculations, top-scoring ligand-enzyme complexes were subjected to 30 ns molecular dynamics simulations. In molecular dynamics simulations, GROMACS [20] program package, AMBER [21] force field and TIP3P water model were used. Acypc Server [22] was used in the preparation of ligand topologies. After energy minimizations, 200 ps NVT and NPT ensemble equilibrations, molecular dynamics simulations were performed for 30 ns at 1 bar and 300 K reference pressure and temperature. After performing molecular dynamics simulations, binding free energies were calculated with the use of MM-PBSA method for the last 20 ns of the MD simulations. In MM-PBSA calculations, g_mmpbsa tool [23,24] was used.

2.3. Drug-likeness analyses

Lipinski's rule of five [25,26] was used to evaluate drug-likeness of the investigated compounds. Drug-likeness of the investigated compounds were tested with the use of DruLiTo software package [27].

3. Results and Discussion

3.1. DFT calculations

Optimized structures of the investigated natural compounds obtained from DFT calculations with the use of 6-311+G(2d,p) basis set are given in Figure 2. A frequency analysis was performed for each structure to confirm that the optimized structures correspond to global minima and results showed that geometry optimized structures correspond to global minima.

To obtain information about the electron deficient and the electron rich regions of the compounds under investigation, MEP map calculations were performed at the same level of theory.

MEP maps of the compounds obtained with the use of 6-311+G(2d,p) basis set are given in Figure 3. It was observed that negative charge was mainly localized on the carbonyl oxygens of both compounds 1 and 2, while positive charge as located generally on the alkyl hydrogens. Molecular docking and molecular dynamics simulation results showed that these negative and positive centers took part in the interactions between investigated compounds and SARS-CoV-2 M^{pro}.

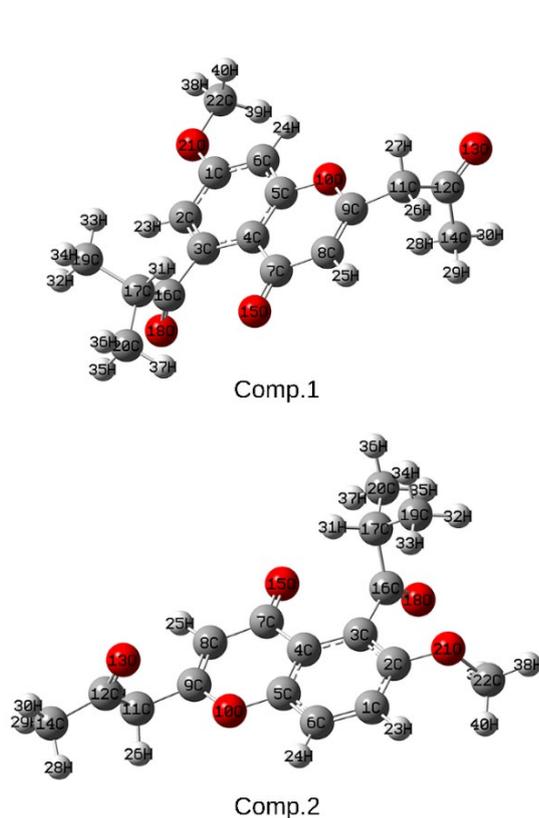


Figure 2. Optimized geometries of Comp. 1 and 2.

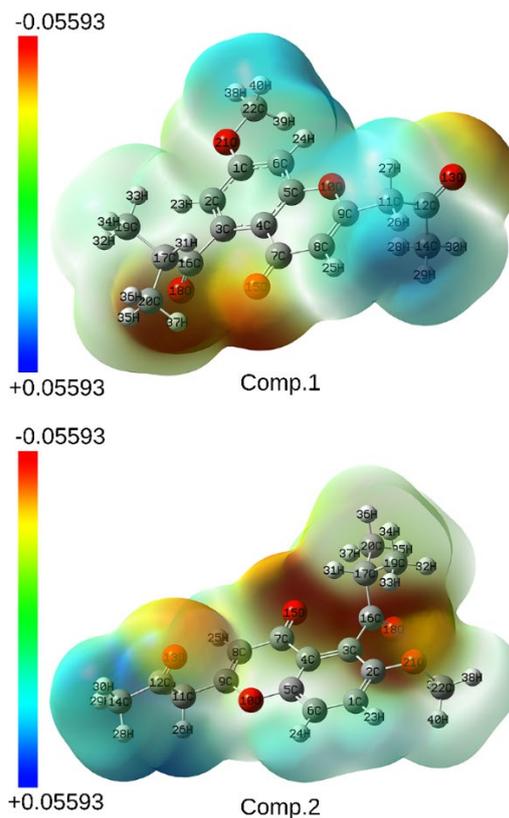


Figure 3. MEP maps of Comp. 1 and 2

FMOs and FMO energies of the investigated compounds have also been determined and are given in Figure 4. It was observed that HOMO-LUMO gap of compound 1 is slightly greater than that of compound 2. HOMO-LUMO gap is thought to be useful parameter for examining the kinetic stability. Since compound 1 has larger HOMO-LUMO gap value, it can be said that compound 1 is more stable than compound 2. HOMO-LUMO energies and HOMO-LUMO gaps of the investigated compounds calculated with the use of 6-31+G(d), 6-31+G(d,p), 6-311+G(d,p) and 6-311+G(2d,p) basis sets are given in Table 1. All values in Table 1 are given in eV.

Nuclear magnetic shield tensors for the compounds were performed at the same level of theory with the use of GIAO method. In this part, 6-31+G(d), 6-31+G(d,p), 6-311+G(d,p) and 6-311+G(2d,p) basis sets were used. In NMR calculations IEFPCM solvation model was used and CDCl_3 was selected as solvent.

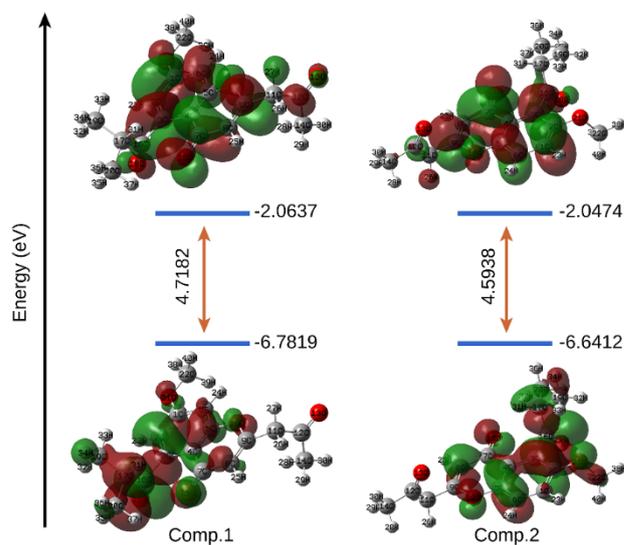


Figure 4. FMOs of Comp. 1 and 2

Calculated and experimental NMR chemical shifts reference to TMS are given in Tables 2, 3, 4 and 5.

Table 1. Calculated FMO energies and HOMO-LUMO gaps of investigated compounds.

Atom	6-31+G(d)	6-31+G(d,p)	6-311+G(d,p)	6-311+G(2d,p)
Comp. 1				
LUMO	-1.7565	-1.7228	-2.0814	-2.0637
HOMO	-6.2488	-6.3928	-6.7735	-6.7819
Gap	4.4923	4.6700	4.6921	4.7182
Comp. 2				
LUMO	-1.7451	-1.6885	-2.0819	-2.0474
HOMO	-6.2390	-6.2864	-6.6682	-6.6412
Gap	4.4939	4.5979	4.5863	4.5938

It was observed that except some certain types of hydrogen and carbon, smaller basis sets gave more satisfactory results than larger basis sets and there is no need to use larger basis sets.

Table 2. Experimental and calculated ¹H-NMR data for Comp. 1

Atom	Exp. [11]	6-31+G(d)	6-31+G(d,p)	6-311+G(d,p)	6-311+G(2d,p)
23-H	7.42	6.72	6.96	6.97	7.08
24-H	7.12	6.74	7.04	7.12	7.23
25-H	6.32	5.95	6.22	6.27	6.37
26-H	3.55	3.56	3.63	3.74	3.83
27-H	3.55	3.56	3.63	3.74	3.83
28-H	2.17	2.17	2.26	2.36	2.40
29-H	2.17	2.17	2.26	2.36	2.40
30-H	2.17	2.17	2.26	2.36	2.40
31-H	4.25	3.05	3.07	3.09	3.22
32-H	1.23	1.03	1.04	1.13	1.13
33-H	1.23	1.03	1.04	1.13	1.13
34-H	1.23	1.03	1.04	1.13	1.13
35-H	1.23	1.46	1.48	1.58	1.56
36-H	1.23	1.46	1.48	1.58	1.56
37-H	1.23	1.46	1.48	1.58	1.56
38-H	3.82	3.92	3.99	4.04	4.06
39-H	3.82	3.92	3.99	4.04	4.06
40-H	3.82	3.92	3.99	4.04	4.06

Table 3. Experimental and calculated ^{13}C -NMR data for Comp. 1

Atom	Exp. [11]	6-31+G(d)	6-31+G(d,p)	6-311+G(d,p)	6-311+G(2d,p)
1-C	166.8	157.8	160.0	172.6	172.6
2-C	106.2	108.8	109.8	119.1	118.7
3-C	136.8	142.8	144.8	154.2	153.9
4-C	114.8	113.0	114.8	123.2	123.2
5-C	155.4	153.9	155.8	166.5	166.8
6-C	110.2	99.5	100.5	108.3	107.7
7-C	181.4	169.8	171.6	183.1	183.8
8-C	112.8	110.7	111.8	120.3	119.9
9-C	158.2	159.5	161.4	172.2	172.5
11-C	48.3	51.0	51.9	55.3	54.9
12-C	203.1	200.1	201.6	215.6	215.6
14-C	30.4	29.9	30.3	32.7	32.4
16-C	208.5	209.4	211.2	226.4	226.0
17-C	38.3	46.2	47.0	49.6	50.0
19-C	18.6	21.3	21.6	21.9	21.5
20-C	18.6	21.3	21.6	21.9	21.5
22-C	56.2	55.0	55.2	58.2	58.1

Table 4. Experimental and calculated ^1H -NMR data for Comp. 2

Atom	Exp. [11]	6-31+G(d)	6-31+G(d,p)	6-311+G(d,p)	6-311+G(2d,p)
23-H	6.93	7.49	7.71	7.74	7.82
24-H	7.10	7.50	7.73	7.79	7.86
25-H	6.35	5.99	6.25	6.31	6.37
26-H	3.56	3.55	3.63	3.76	3.82
27-H	3.56	3.55	3.63	3.76	3.82
28-H	2.17	2.18	2.26	2.38	2.40
29-H	2.17	2.18	2.26	2.38	2.40
30-H	2.17	2.18	2.26	2.38	2.40
31-H	4.27	3.03	3.05	3.09	3.16
32-H	1.26	1.30	1.31	1.42	1.37
33-H	1.26	1.30	1.31	1.42	1.37
34-H	1.26	1.30	1.31	1.42	1.37
35-H	1.26	1.30	1.31	1.42	1.37
36-H	1.26	1.30	1.31	1.42	1.37
37-H	1.26	1.30	1.31	1.42	1.37
38-H	3.84	3.82	3.91	3.97	4.02
39-H	3.84	3.82	3.91	3.97	4.02
40-H	3.84	3.82	3.91	3.97	4.02

Table 5. Experimental and calculated ^{13}C -NMR data for Comp. 2

Atom	Exp. [11]	6-31+G(d)	6-31+G(d,p)	6-311+G(d,p)	6-311+G(2d,p)
1-C	120.2	126.2	127.3	137.6	137.1
2-C	156.4	150.6	152.6	164.3	164.0
3-C	121.4	135.2	137.0	146.6	146.4
4-C	118.5	120.0	121.8	130.7	130.9
5-C	150.4	149.4	151.4	161.6	161.8
6-C	122.9	116.1	117.3	126.7	126.1
7-C	181.8	170.5	172.3	183.8	184.5
8-C	110.9	109.9	111.0	119.6	119.2
9-C	159.9	159.9	161.8	172.4	172.9
11-C	48.1	51.7	52.5	56.0	55.6
12-C	203.4	199.9	201.4	215.1	215.1
14-C	30.2	29.8	30.1	32.6	32.2
16-C	208.4	208.4	210.2	224	223.7
17-C	38.2	47.6	48.4	50.6	50.8
19-C	18.6	21.0	21.3	21.6	21.2
20-C	18.6	21.0	21.3	21.6	21.2
22-C	56.2	66.2	66.6	69.1	69.4

3.2. Molecular docking calculations and molecular dynamics simulation studies

After performing molecular docking calculations, 30 ns molecular dynamics simulations were carried out.

The structures of the SARS-CoV-2 M^{pro} – Comp. 1 complex obtained from the trajectory for every 5 ns of the 30 ns molecular dynamics simulation are given in Figure 5.

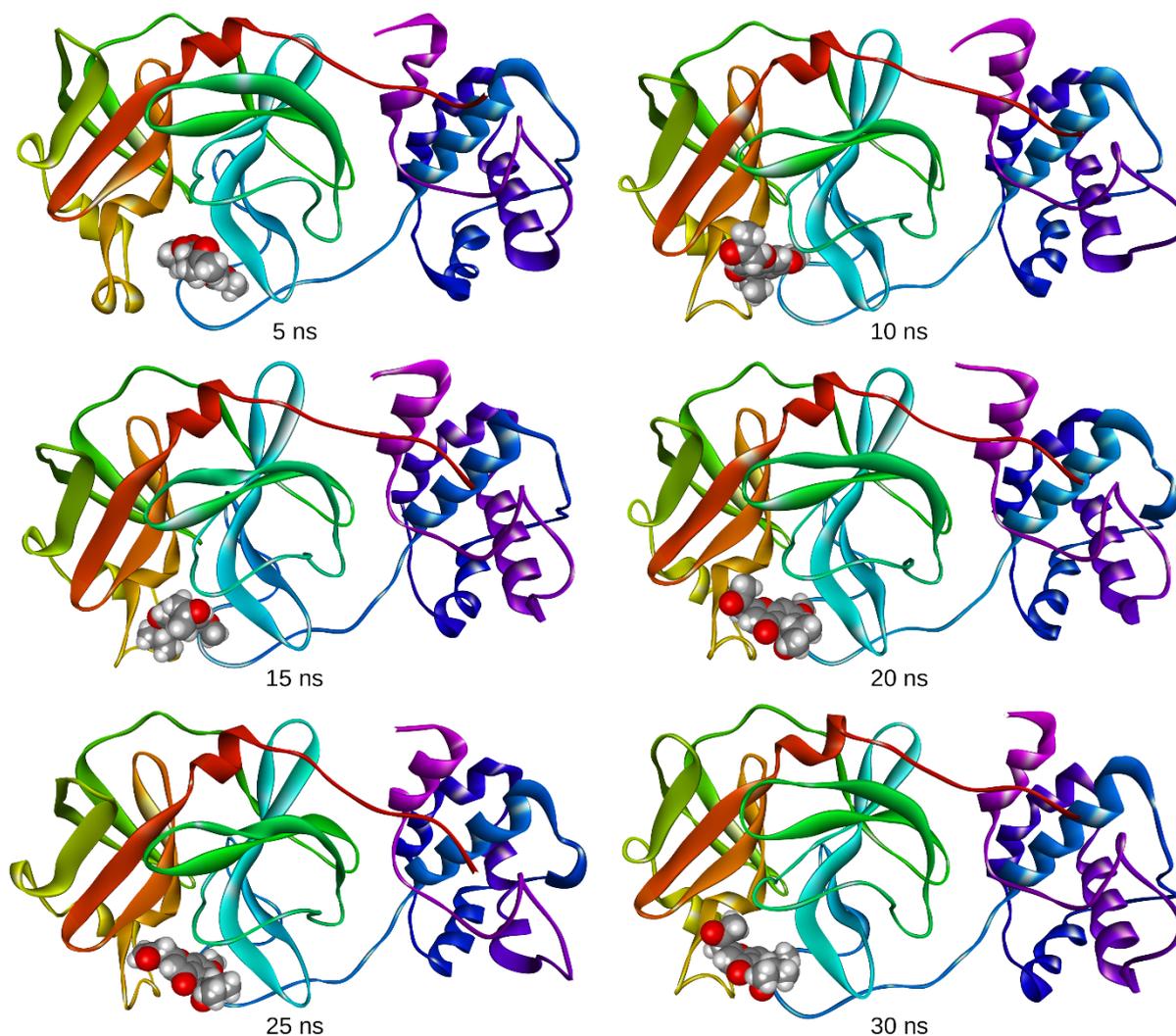


Figure 5. The structures of enzyme – Comp. 1 complex extracted from the trajectory of MD simulation.

Results showed that Comp. 1 bound to the active site of SARS-CoV-2 M^{pro} and held its position throughout the entire simulation. The structures of the SARS-CoV-2 M^{pro} – Comp. 2 complex obtained from the trajectory for every 5 ns of the 30 ns molecular

dynamics simulation are given in Figure 6 and it was observed that Comp. 2 also bound to the active site of the enzyme and held its position throughout the entire simulation.

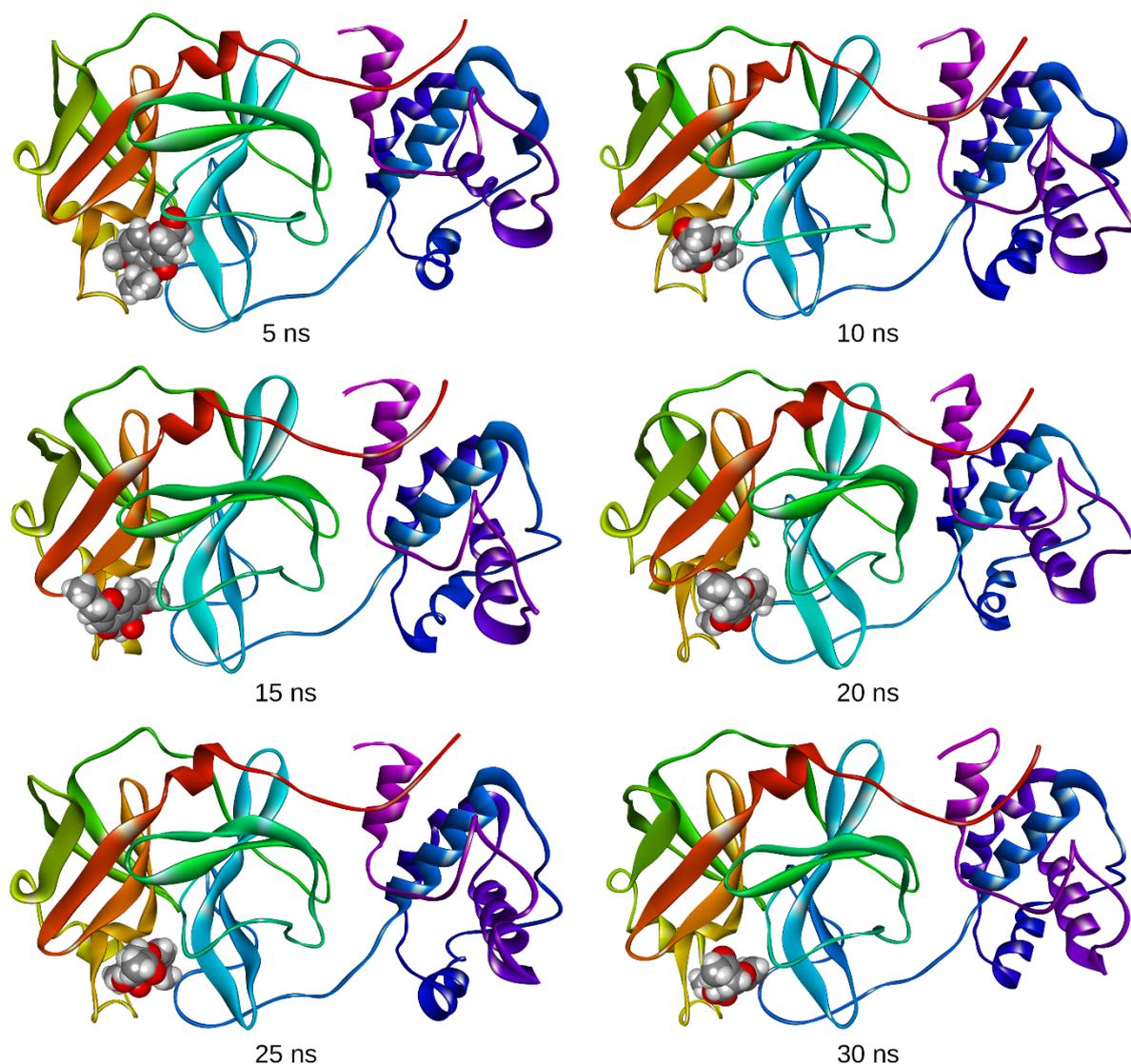


Figure 6. The structures of enzyme – Comp. 2 complex extracted from the trajectory of MD simulation.

In Figure 7, 3D and 2D interactions between investigated compounds and SARS-CoV-2 M^{pro} at the end of molecular dynamics simulations are illustrated. Results showed that, Comp. 1 interacted with HIS41, MET49, MET165, VAL186 and GLN189 amino acids of SARS-CoV-2 M^{pro}. Hydrogen bonds, alkyl, π -alkyl

and π -sulfur interactions took part in the stabilization of the enzyme – Comp. 1 complex. On the other hand, Comp. 2 interacted with HIS41, MET49, GLY143, CYS145 and SER46. Hydrogen bonds, alkyl and π -alkyl interactions took part in the stabilization of enzyme – Comp. 2 complex.

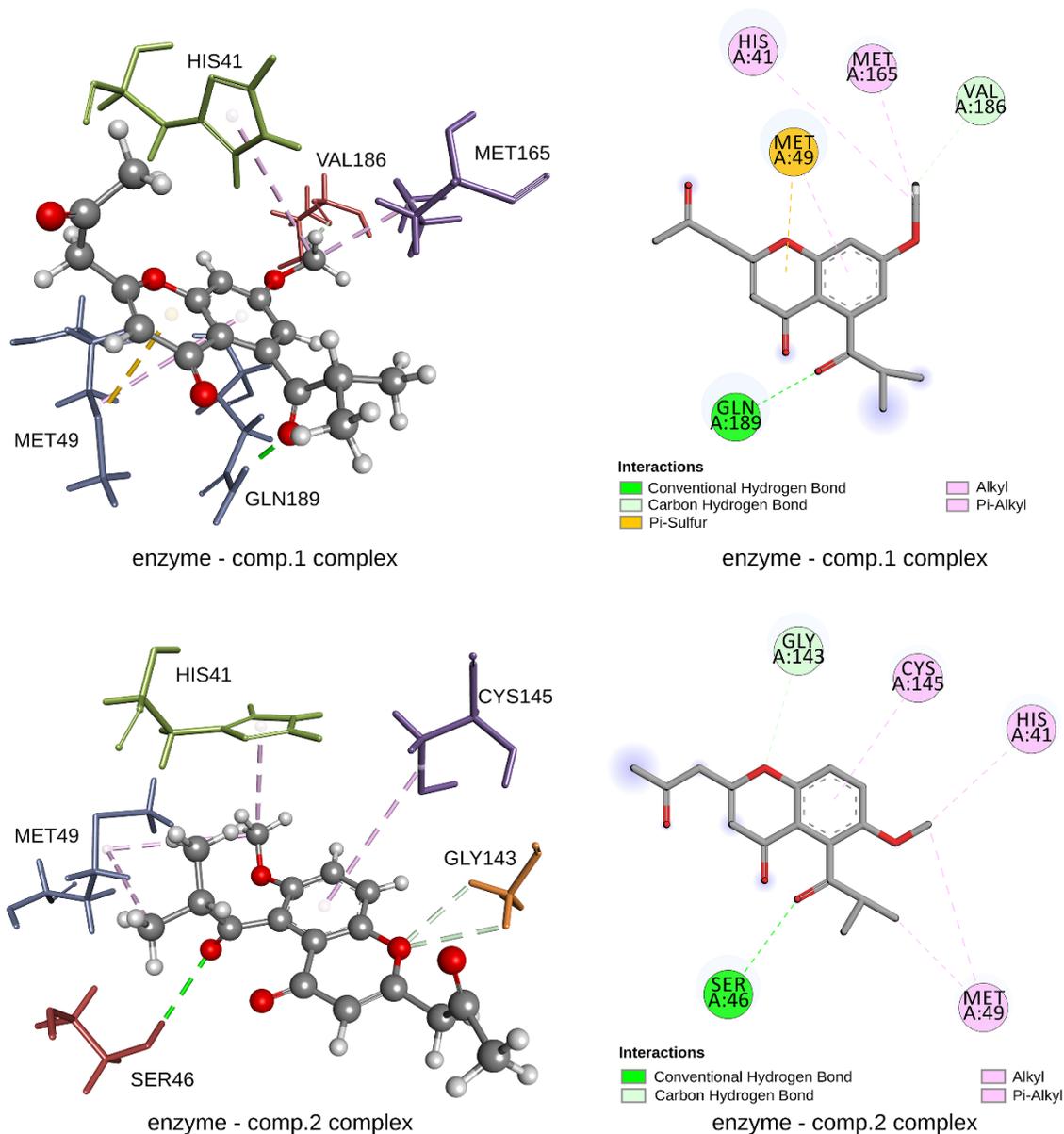


Figure 7. Interactions between ligands and enzyme after MD simulation.

Molecular dynamics simulation results for enzyme – Comp. 1 complex are illustrated in Figure 8. Root mean square deviation (RMSD) and radius of gyration (RG) of protein are useful tools for determining the stability of a ligand-protein complex. Results showed that enzyme – Comp. 1 complex remained stable during the MD simulation (Figure 8b and 8c). Average RMSD of backbone after least square fit to backbone was found to be 0.189 ± 0.021 nm and average RG of protein was found to be 2.224 ± 0.014 nm. It was observed that there is a noticeable change in the

position of Comp. 1 around 7 ns and then, a slight change around 18 ns. Results showed that ligand reached its equilibrium position at the 18th ns of the simulation and held its position for the remaining time of the simulation (Figure 8a). Average RMSD of Comp. 1 after least square fit to protein was found to be 0.499 ± 0.152 nm. Number of hydrogen bonds was also monitored during the MD simulation (Figure 8d). Results showed that at least one hydrogen bond was formed between the ligand and the enzyme in nearly one third of the simulation time.

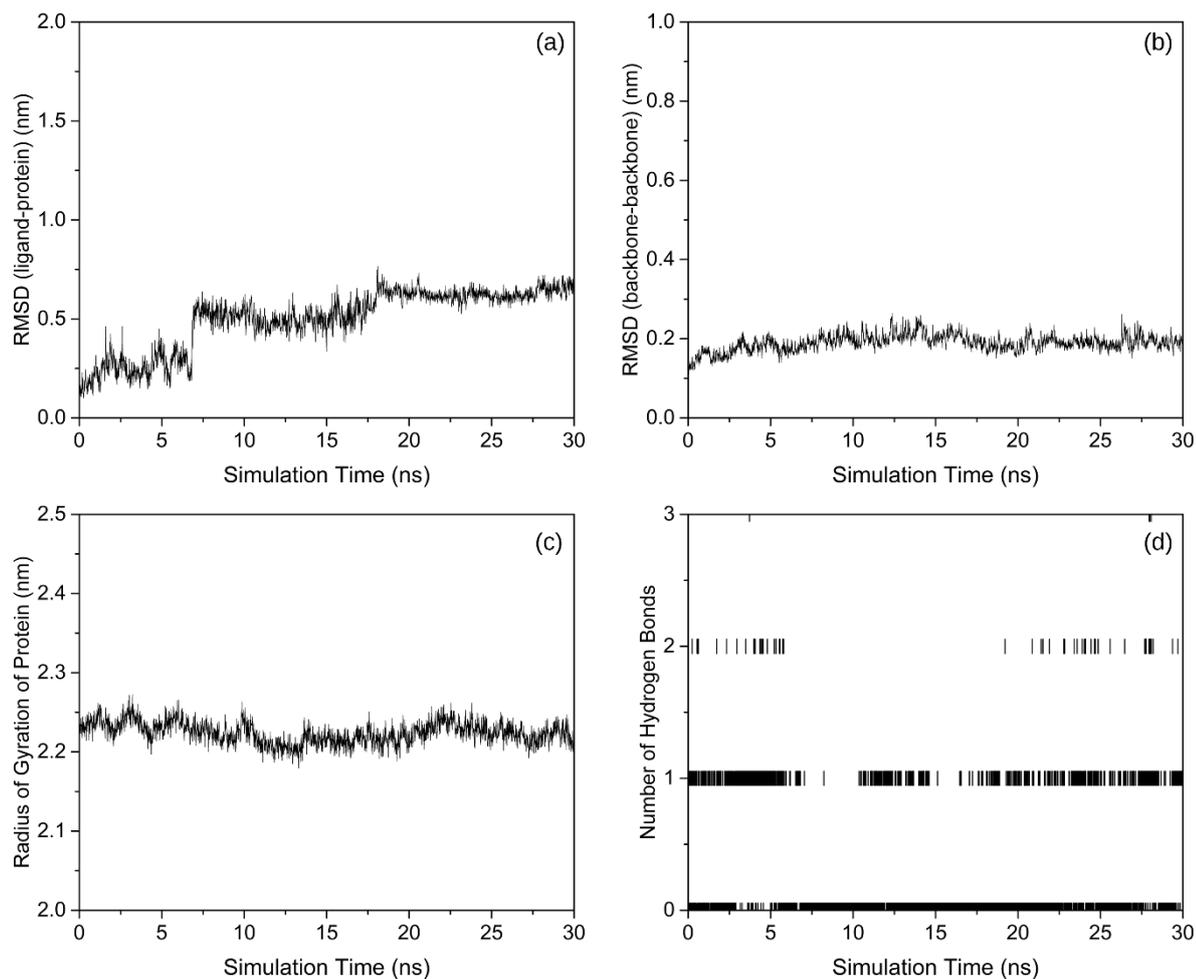


Figure 8. MD simulation results (a) RMSD of ligand after least square fit to protein, (b) RMSD of backbone after least square fit to backbone, (c) Radius of gyration of protein and (d) number of hydrogen bonds between protein and Comp. 1.

Molecular dynamics simulation results for enzyme – Comp. 2 complex are illustrated in Figure 9. Results showed that enzyme remained stable during the MD simulation (Figure 9b and 9c). Average RMSD of backbone after least square fit to backbone was found to be 0.219 ± 0.031 nm and average RG of protein was found to be 2.212 ± 0.010 nm. It was observed that RMSD of ligand after least square fit to protein

increased until 17th ns, and then it decreased gradually and reached its equilibrium position at the 24th ns of the MD simulation (Figure 9a). Average RMSD of Comp. 2 after least square fit to protein was found to be 0.612 ± 0.173 nm. Results showed that at least one hydrogen bond was formed between Comp. 2 and the enzyme in nearly half of the simulation time.

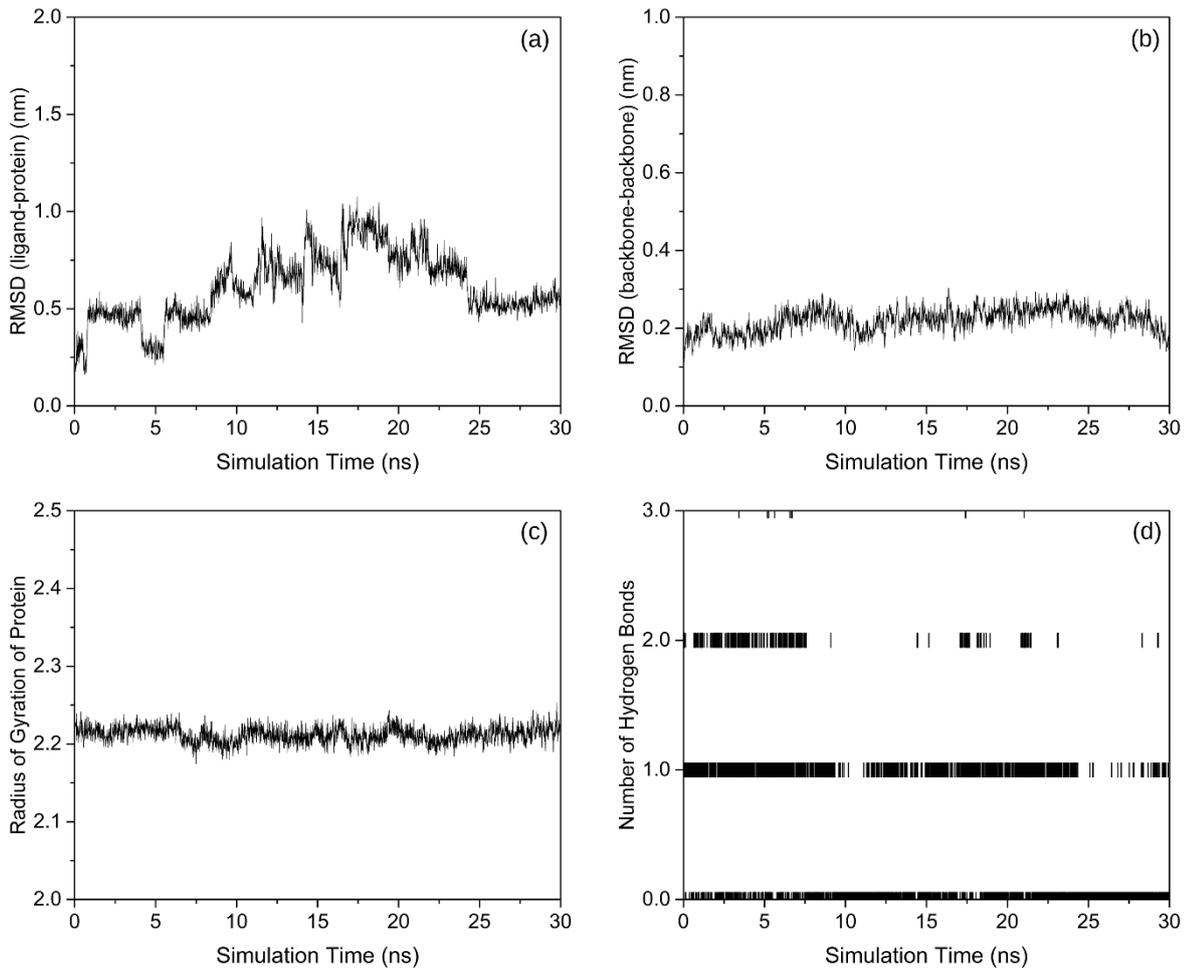


Figure 9. MD simulation results (a) RMSD of ligand after least square fit to protein, (b) RMSD of backbone after least square fit to backbone, (c) Radius of gyration of protein and (d) number of hydrogen bonds between protein and Comp. 2.

Binding free energies were calculated with the use of MM-PBSA method (Figure 10). Binding free energies of the reference drugs are literature values which were determined in our previous work. [8] Results showed that binding affinities of the investigated compounds

are quite high. It was observed that although the binding affinity of Comp. 1 is lower than those of lopinavir and hydroxychloroquine, it was higher than that of remdesivir.

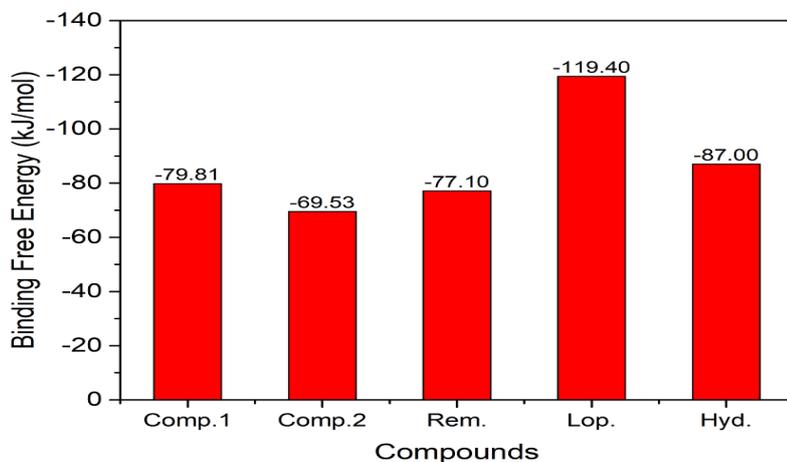


Figure10. Binding free energies belong to investigated compounds and reference drugs obtained from MM-PBSA calculations [8] (Rem.: Remdesivir, Lop.:Lopinavir and Hyd.: Hydroxychloroquine).

3.3. Drug-likeness analysis

Lipinski's rule of five [25,26] was used to evaluate drug-likeness of the investigated compounds. Drug-likeness of the investigated compounds were tested with the use of DruLiTo software package [27]. As known, Lipinski's rule describes the molecular properties which are important for a drug's pharmacokinetics, including absorption, distribution, metabolism, and excretion. According to Lipinski's rule a potential drug molecule has no more than one violation of the following criteria:

- A molecular mass less than 500 daltons
- An octanol-water partition coefficient (logP) that doesn't exceed 5
- No more than 5 HBD (hydrogen bond donors)
- No more than 10 HBA (hydrogen bond acceptors)

For the investigated molecules (both Comp. 1 and 2), molecular weight, logP, HBD and HBA values were found to be 302.12, 0.999, 0 and 5, respectively (Table 6). Results showed that there is no violation of the criteria of Lipinski's rule.

Table 6. Results of drug-likeness analyses.

Parameter	Criteria	Calculated Value	
		Comp. 1	Comp. 2
MW	< 500	302.12	302.12
logP	< 5	0.999	0.999
HBD	< 5	0	0
HBA	< 10	5	5

4. Conclusion

In the present study, two newly introduced chromone derivatives have been investigated for their possible antiviral activity against SARS-CoV-2 with the assistance of computational methods including DFT calculations, molecular docking calculations and molecular dynamics simulations. In the study, MM-PBSA calculations and drug-likeness analyses were also carried out. Results showed that investigated compounds efficiently bound to SARS-CoV-2 M^{Pro} and held their positions throughout the entire molecular dynamics simulations. Binding affinities of the investigated compounds were found to be quite high and comparable to the reference drugs. Especially binding affinity of Comp. 1 is higher than that of Comp. 2 and reference drug remdesivir. However, it was found that the binding affinities of the investigated

compounds are lower than those of lopinavir and hydroxychloroquine. Additionally, it was observed that both compounds obey Lipinski's rule of five and no violation was observed. In the study it was concluded that these natural chromone derivatives can be promising structures in the treatment of SARS-CoV-2 infections and worth for further research.

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Conflict of interest

The authors declare that they have no conflict of interests.

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